

PATENT COOPERATION TREATY

Plougmann & Vingtoft

From the
INTERNATIONAL PRELIMINARY EXAMINING AUTHORITY

06 JUNI 2006

PCT PJN/SPK

To:

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NOTIFICATION OF TRANSMITTAL OF
THE INTERNATIONAL PRELIMINARY
REPORT ON PATENTABILITY
(PCT Rule 71.1)

Date of mailing
(day/month/year)

01.06.2006

Applicant's or agent's file reference
36029PC02

IMPORTANT NOTIFICATION

International application No.
PCT/DK2004/000888

International filing date (day/month/year)
19.12.2004

Priority date (day/month/year)
19.12.2003

Applicant
SCF TECHNOLOGIES A/S

1. The applicant is hereby notified that this International Preliminary Examining Authority transmits herewith the international preliminary report on patentability and its annexes, if any, established on the international application.
2. A copy of the report and its annexes, if any, is being transmitted to the International Bureau for communication to all the elected Offices.
3. Where required by any of the elected Offices, the International Bureau will prepare an English translation of the report (but not of any annexes) and will transmit such translation to those Offices.
4. **REMINDER**

The applicant must enter the national phase before each elected Office by performing certain acts (filing translations and paying national fees) within 30 months from the priority date (or later in some Offices) (Article 39(1)) (see also the reminder sent by the International Bureau with Form PCT/IB/301).

Where a translation of the international application must be furnished to an elected Office, that translation must contain a translation of any annexes to the international preliminary report on patentability. It is the applicant's responsibility to prepare and furnish such translation directly to each elected Office concerned.

For further details on the applicable time limits and requirements of the elected Offices, see Volume II of the PCT Applicant's Guide.

The applicant's attention is drawn to Article 33(5), which provides that the criteria of novelty, inventive step and industrial applicability described in Article 33(2) to (4) merely serve the purposes of international preliminary examination and that "any Contracting State may apply additional or different criteria for the purposes of deciding whether, in that State, the claimed inventions is patentable or not" (see also Article 27(5)). Such additional criteria may relate, for example, to exemptions from patentability, requirements for enabling disclosure, clarity and support for the claims.

Name and mailing address of the international
preliminary examining authority:



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PATENT COOPERATION TREATY

PCT

INTERNATIONAL PRELIMINARY REPORT ON PATENTABILITY

(Chapter II of the Patent Cooperation Treaty)

(PCT Article 36 and Rule 70)

Applicant's or agent's file reference 36029PC02		FOR FURTHER ACTION		See Form PCT/PEAA16
International application No. PCT/DK2004/000888		International filing date (day/month/year) 19.12.2004		Priority date (day/month/year) 19.12.2003
International Patent Classification (IPC) or national classification and IPC INV. B01J204				
Applicant SCF TECHNOLOGIES AS				
<p>1. This report is the international preliminary examination report, established by this International Preliminary Examining Authority under Article 35 and transmitted to the applicant according to Article 36.</p> <p>2. This REPORT consists of a total of 9 sheets, including this cover sheet.</p> <p>3. This report is also accompanied by ANNEXES, comprising:</p> <p>a. <input checked="" type="checkbox"/> sent to the applicant and to the International Bureau) a total of 16 sheets, as follows:</p> <p style="margin-left: 40px;"><input checked="" type="checkbox"/> sheets of the description, claims and/or drawings which have been amended and are the basis of this report and/or sheets containing rectifications authorized by this Authority (see Rule 70.16 and Section 607 of the Administrative Instructions).</p> <p style="margin-left: 40px;"><input type="checkbox"/> sheets which supersede earlier sheets, but which this Authority considers contain an amendment that goes beyond the disclosure in the international application as filed, as indicated in item 4 of Box No. I and the Supplemental Box.</p> <p>b. <input type="checkbox"/> (sent to the International Bureau only) a total of (indicate type and number of electronic carrier(s)) , containing a sequence listing and/or tables related thereto, in electronic form only, as indicated in the Supplemental Box Relating to Sequence Listing (see Section 802 of the Administrative Instructions).</p>				
<p>4. This report contains indications relating to the following items:</p> <p><input checked="" type="checkbox"/> Box No. I Basis of the report</p> <p><input type="checkbox"/> Box No. II Priority</p> <p><input checked="" type="checkbox"/> Box No. III Non-establishment of opinion with regard to novelty, inventive step and industrial applicability</p> <p><input checked="" type="checkbox"/> Box No. IV Lack of unity of invention</p> <p><input checked="" type="checkbox"/> Box No. V Reasoned statement under Article 35(2) with regard to novelty, inventive step or industrial applicability; citations and explanations supporting such statement</p> <p><input type="checkbox"/> Box No. VI Certain documents cited</p> <p><input type="checkbox"/> Box No. VII Certain defects in the international application</p> <p><input type="checkbox"/> Box No. VIII Certain observations on the international application</p>				
Date of submission of the demand 02.12.2005			Date of completion of this report 01.06.2006	
Name and mailing address of the international preliminary examining authority:  European Patent Office - Gitschiner Str. 103 D-10958 Berlin Tel. +49 30 25901 - 0 Fax: +49 30 25901 - 840			Authorized officer Clement, J-P Telephone No. +49 30 25901-325	



INTERNATIONAL PRELIMINARY REPORT ON PATENTABILITY

International application No.
PCT/DK2004/000888

Box No. I Basis of the report

1. With regard to the language, this report is based on

- ☒ the international application in the language in which it was filed
- ☐ a translation of the international application into , which is the language of a translation furnished for the purposes of:
 - ☐ international search (under Rules 12.3(a) and 23.1(b))
 - ☐ publication of the international application (under Rule 12.4(a))
 - ☐ international preliminary examination (under Rules 55.2(a) and/or 55.3(a))

2. With regard to the **elements*** of the international application, this report is based on *(replacement sheets which have been furnished to the receiving Office in response to an invitation under Article 14 are referred to in this report as "originally filed" and are not annexed to this report):*

Description, Pages

1-30 as originally filed

Claims, Numbers

1-147 filed with telefax on 05.05.2006

Drawings, Sheets

1/7-7/7 as originally filed

- ☐ a sequence listing and/or any related table(s) - see Supplemental Box Relating to Sequence Listing

3. ☐ The amendments have resulted in the cancellation of:

- ☐ the description, pages
- ☐ the claims, Nos.
- ☐ the drawings, sheets/figs
- ☐ the sequence listing (*specify*):
- ☐ any table(s) related to sequence listing (*specify*):

4. ☐ This report has been established as if (some of) the amendments annexed to this report and listed below had not been made, since they have been considered to go beyond the disclosure as filed, as indicated in the Supplemental Box (Rule 70.2(c)).

- ☐ the description, pages
- ☐ the claims, Nos.
- ☐ the drawings, sheets/figs
- ☐ the sequence listing (*specify*):
- ☐ any table(s) related to sequence listing (*specify*):

* If item 4 applies, some or all of these sheets may be marked "superseded."

**INTERNATIONAL PRELIMINARY REPORT
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Box No. III Non-establishment of opinion with regard to novelty, inventive step and industrial applicability

1. The questions whether the claimed invention appears to be novel, to involve an inventive step (to be non-obvious), or to be industrially applicable have not been examined in respect of:

- ☐ the entire international application,
☒ claims Nos. 139-142

because:

- ☐ the said international application, or the said claims Nos. relate to the following subject matter which does not require an international preliminary examination (*specify*):
- ☐ the description, claims or drawings (*indicate particular elements below*) or said claims Nos. are so unclear that no meaningful opinion could be formed (*specify*):
- ☐ the claims, or said claims Nos. are so inadequately supported by the description that no meaningful opinion could be formed (*specify*).
- ☒ no international search report has been established for the said claims Nos. 139-142
- ☐ a meaningful opinion could not be formed without the sequence listing; the applicant did not, within the prescribed time limit:
- ☐ furnish a sequence listing on paper complying with the standard provided for in Annex C of the Administrative Instructions, and such listing was not available to the International Preliminary Examining Authority in a form and manner acceptable to it.
- ☐ furnish a sequence listing in electronic form complying with the standard provided for in Annex C of the Administrative Instructions, and such listing was not available to the International Preliminary Examining Authority in a form and manner acceptable to it.
- ☐ pay the required late furnishing fee for the furnishing of a sequence listing in response to an invitation under Rules 13*ter*.1(a) or (b) and 13*ter*.2.
- ☐ a meaningful opinion could not be formed without the tables related to the sequence listings; the applicant did not, within the prescribed time limit, furnish such tables in electronic form complying with the technical requirements provided for in Annex C-*bis* of the Administrative Instructions, and such tables were not available to the International Preliminary Examining Authority in a form and manner acceptable to it.
- ☐ the tables related to the nucleotide and/or amino acid sequence listing, if in electronic form only, do not comply with the technical requirements provided for in Annex C-*bis* of the Administrative Instructions.
- ☐ See separate sheet for further details

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Box No. IV Lack of unity of invention

1. ☒ In response to the invitation to restrict or pay additional fees, the applicant has, within the applicable time limit:
- ☐ restricted the claims.
 - ☐ paid additional fees.
 - ☐ paid additional fees under protest and, where applicable, the protest fee.
 - ☐ paid additional fees under protest but the applicable protest fee was not paid.
 - ☒ neither restricted the claims nor paid additional fees.
2. ☐ This Authority found that the requirement of unity of invention is not complied with and chose, according to Rule 68.1, not to invite the applicant to restrict or pay additional fees.
3. This Authority considers that the requirement of unity of invention in accordance with Rules 13.1, 13.2 and 13.3 is:
- ☐ complied with.
 - ☒ not complied with for the following reasons:
see separate sheet
4. Consequently, this report has been established in respect of the following parts of the international application:
- ☐ all parts.
 - ☒ the parts relating to claims Nos. 1-138, 143-147 .

Box No. V Reasoned statement under Article 35(2) with regard to novelty, inventive step or industrial applicability; citations and explanations supporting such statement

1. Statement

Novelty (N)	Yes: Claims	1-138, 143-147
	No: Claims	
Inventive step (IS)	Yes: Claims	1-16
	No: Claims	17-138, 143-147
Industrial applicability (IA)	Yes: Claims	1-138, 143-147
	No: Claims	

2. Citations and explanations (Rule 70.7):

see separate sheet

**INTERNATIONAL PRELIMINARY REPORT
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**INTERNATIONAL PRELIMINARY
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International application No.

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Re Item III

Due to the objection of non unity (see Item IV below), claims 139-142 were not searched and not examined.

Re Item IV.

The separate inventions/groups of inventions are:

Claims 1-138, 143-147

a method of producing a fine particle comprising introducing one or more substances contained in a supercritical fluid into a vessel comprising a material, causing said substances to precipitate as primary particles on the surface of said material and a product obtainable according to the method

Claims 139-142

a tape cast comprising particles deposited on a carrier film, wherein said primary particles have an average diameter of less than 100 nanometer, and a narrow size distribution around the average diameter characterized by having a maximum standard deviation of said distribution of maximum 20 nanometer

They are not so linked as to form a single general inventive concept (Rule 13.1 PCT) for the following reasons:

The inventions listed above do not relate to a single inventive concept under PCT Rule 13.1 because, under PCT Rule 13.2, they lack the same or corresponding special technical features for the following reasons: The special technical feature which defines the contribution of invention I over the prior art is, according to the applicant, the presence of a material in one or more sections of the vessel and the precipitation of the substance on the surface of said material. The special technical feature of invention II is the presence of primary particles on a carrier film, wherein said primary particles have an average diameter of less than 100 nanometer and a maximum standard deviation of the size distribution around the average diameter of maximum 20 nanometer. The single common concept linking the two inventions is the coating of a material with fine particles. This concept is not novel: see document US-A-5789027; It is clear that the special technical

features of inventions I and II are not related as to form a single inventive concept. Due to the fact that no other technical features can be regarded as special technical feature in the sense of rule 13.2 PCT, the ISA is of the opinion that there is no single inventive concept underlying the 2 inventions in the sense of rule 13.1 PCT.

Re Item V

**Reasoned statement with regard to novelty, inventive step or industrial applicability;
citations and explanations supporting such statement**

The amendments filed with the International Bureau under Article 19(1) are admissible as they do not introduce subject-matter which extends beyond the content of the application as filed, according to Article 19(2) PCT.

Reference is made to the following document/s/:

D1: US-A-5789027

D2: US-A-6056791

The document D1 is regarded as being the closest prior art to the subject-matter of The document D1 discloses(col 1, line53, col 7, line 62):

a method for depositing a film of material on the surface of a substrate by dissolving a precursor of the material into a superficial supercritical solvent to form a supercritical solution; exposing the substrate to the solution; mixing a reaction reagent into the solution under conditions that initiate a chemical reaction involving the precursor, thereby depositing the material onto the substrate.

The subject-matter of claim 1 differs from this known D1 in that it comprises the subsequent step of introducing into the vessel one or more substances dissolved and/or dispersed in at least one fluid

The subject-matter of claim 1 is therefore new (Article 33(2) PCT).

By subsequently introducing the substances dissolved and/or dispersed in at least one fluid, the chemical reaction occurs on the surface of the material and not with the reactants in the fluid, since this has already happen in the first step of introducing the substance. Consequently the method has a more controllable growth.

The subject-matter of claim 1 is therefore considered as involving an inventive step (Article 33(3)).

Claims 2-16 are dependent on claim 1 and as such also meet the requirements of the PCT with respect to novelty and inventive step.

The present application does not meet the criteria of Article 33(1) PCT, because the subject-matter of independent claims 17-18 does not involve an inventive step in the sense of Article 33(3) PCT.

The subject-matter of claims 16 and 17 differs from the process in D1 in that:

- * the precipitation is provided/caused by a change in the solubility of at least one of said substances ,
 - * wherein an antisolvent is one of the fluids being introduced to the vessel (claim 17) or *
- said change of solubility is provided/caused by expanding one of said fluids containing at least one of said substances into the vessel.

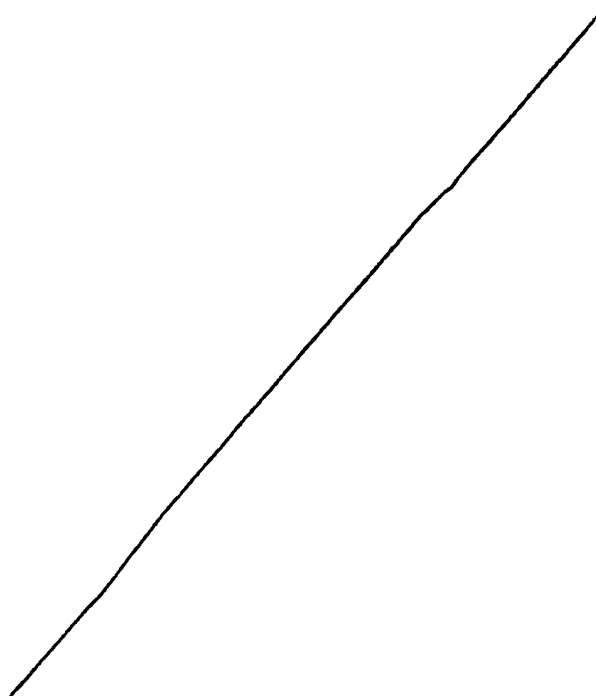
However all those features are well established technics in the field of powder production in supercritical fluids (see D2 (page 1)) not giving rise to any unexpected technical effect.

Dependent claims 19-138, 143-147 do not contain any features which, in combination with the features of any claim to which they refer, meet the requirements of the PCT in respect of novelty and/or inventive step. All those features relate to alternative design or method , not giving rise to any unexpected technical effect.

**INTERNATIONAL PRELIMINARY
REPORT ON PATENTABILITY
(SEPARATE SHEET)**

International application No.

PCT/DK2004/000888



10/583024

AP20 Rec'd PCT/PTO 15 JUN 2006

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CLAIMS

1. A method of producing a fine particle material comprising
 - introducing one or more substances contained, such as dissolved and/or
5 dispersed in one or more fluid(s) into a vessel by introducing said fluid(s) into the vessel, said vessel containing one or more section(s) comprising a material, at least one of the fluids being in a supercritical state before or after being introduced into said vessel,
 - causing and/or allowing, said substances to precipitate at least partly as primary
10 particles on the surface of said material,
wherein at least one of said substances undergoes a chemical reaction,
wherein at least one said substances undergoing a chemical reaction is an alkoxide, and
wherein said alkoxide comprises a metal- or semi-metal alkoxide,
wherein the method further comprises:
 - 15 i. introducing into the vessel at least one of reactant(s) and/or precursor(s) and/or initiator(s) and/or catalyst(s) for said chemical reaction
 - 20 ii. subsequently introducing into the vessel said one or more substances dissolved and/or dispersed in at least one fluid.
 - iii. and vice versa.
2. A method according to claim 19, wherein said chemical reaction(s) is/are a sol-gel
25 reaction(s).
3. A method according to claim 1 or 2, wherein the average diameter of said
nanoscaled primary particles is smaller than 100 nanometer such as smaller than 30
nanometer, preferably smaller than 20 nanometer, and even more preferable below 15
nanometer such as below 10 nanometer
30
4. A method according to any of the claims 1-3, wherein the standard deviation of the
size distribution of said primary particles formed is less than 60 % of the average diameter
of said primary particles, such as 40 % of the average diameter of said primary particles,
and preferable less than 30 % of the average size of said primary particles, and even more
35 preferable less than 20 % of the average size of said primary particles such as less than 15
% of the average size of said primary particles.

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5. A method according to any of the claims 1-4, wherein the standard deviation of the size distribution of said primary particles formed is maximum 20 nanometer, such as maximum 10 nanometer, and preferably less than 5 nanometer, and even more preferably less than 3 nanometer.

5

6. A method according to any of the claims 1-5, wherein at least one of said fluid(s) being in a supercritical state is selected from the group consisting of carbon dioxide, alcohols such as methanol, ethanol, propanol, isopropanol, butanol, isobutanol, pentanol, hexanol, water, methane, ethane, propane, butane, pentane, hexane, 10 cyclohexane, heptane, ammonia, sulfurhexafluoride, nitrous oxide, chlorotrifluoromethane, monofluoromethane, acetone, THF, acetic acid, citric acid, ethylene glycol, polyethylene glycol, N,N-dimethylaniline and mixtures thereof

15

7. A method according to claim 6, wherein one of the fluids is CO₂.

8. A method according to claim 6, wherein one of the fluids is an organic solvent.

9. A method according to any of claim 6, wherein one of the fluids is water.

20 10. A method according to any of the preceding claims, wherein the pressure of at least one of said fluids is in the range 85-500 bar, preferably in the range 100-300 bar, such as in the range 100-300 bar.

25 11. A method according to any of the preceding claims, wherein the temperature in the vessel is maintained in the range 20-500 °C, such as 30-450 °C, and preferable in the range 35-200 °C, and more preferable in the range 40-150 °C.

12. A method according to any of the preceding claims, wherein said fluid further comprises at least one co-solvent.

30

13. A method according to claim 12, wherein the co-solvent is selected from the group consisting of alcohol(s), water, ethane, ethylene, propane, butane, pentane, hexane, heptane, ammonia, sulfurhexafluoride, nitrous oxide, chlorotrifluoromethane, monofluoromethane, methanol, ethanol, propanol, isopropanol, butanol, pentanol, 35 hexanol, acetone, DMSO, THF, acetic acid, ethyleneglycol, polyethyleneglycol, N,N-dimethylaniline and mixtures thereof.

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14. A method according to any of the preceding claims, wherein fluid further comprises one or more surfactants, said surfactants being preferably selected from the group consisting of hydrocarbons and fluorocarbons preferably having a hydrophilic/lipophilic balance value of less than 15, where the HLB value is determined according to the following formula: $HLB = 7 + \text{sum}(\text{hydrophilic group numbers}) - \text{sum}(\text{lipophilic group numbers})$.
15. A method according to any of the claims 1-14, wherein said precipitation is provided/caused by a change in the solubility of at least one of said substances.
16. A method according to claim 15, wherein said change in the solubility is provided/caused by an antisolvent present in the vessel.
17. A method of producing a fine particle material comprising
- 15 - introducing one or more substances contained, such as dissolved and/or dispersed in one or more fluid(s) into a vessel by introducing said fluid(s) into the vessel, said vessel containing one or more section(s) comprising a material, at least one of the fluids being in a supercritical state before or after being introduced into said vessel,
- 20 - causing and/or allowing, said substances to precipitate at least partly as primary particles on the surface of said material,
- wherein said precipitation is provided/caused by a change in the solubility of at least one of said substances, and wherein said antisolvent is one of the fluids being introduced to the vessel.
- 25
18. A method of producing a fine particle material comprising
- introducing one or more substances contained, such as dissolved and/or dispersed in one or more fluid(s) into a vessel by introducing said fluid(s) into the vessel, said vessel containing one or more section(s) comprising a material, at least one of the fluids being in a supercritical state before or after being introduced into said vessel,
- 30 - causing and/or allowing, said substances to precipitate at least partly as primary particles on the surface of said material,
- wherein said precipitation is provided/caused by a change in the solubility of at least one of said substances, and wherein said change in solubility of at least one of said substances is provided/caused by expanding at least one of said fluids containing at least one of said substances into the vessel.
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19. A method according to any of the claims 15-18, wherein said change in solubility is provided/caused by a change in the temperature of said fluid(s).
20. A method according to any of claims 16-17, wherein at least one of said substances
5 undergoes a chemical reaction.
21. A method according to claim 20, wherein said chemical reaction(s) is/are a sol-gel reaction(s).
- 10 22. A method according to claim 21, wherein the maximum temperature in the vessel during said sol-gel reaction(s) is maintained below 400 C, such as below 300 C, preferably below 250 C such as below 200 C, and even more preferably below 150 C such as below 100 C.
- 15 23. A method according to claims 20-22, wherein at least one said substances undergoing a chemical reaction is a sulphate salt or a halide such as a chloride.
24. A method according to claims 20-22, wherein at least one said substances undergoing a chemical reaction is an alkoxide.
20
25. A method according to claim 24, wherein said alkoxide comprises a metal- or semi-metal alkoxide.
26. A method according to claim 25 comprising:
25
- iv. introducing into the vessel at least one of said reactant(s) and/or precursor(s) and/or initiator(s) and/or catalyst(s) for said chemical reaction
- 30 v. subsequently introducing into the vessel one or more substances dissolved and/or dispersed in at least one fluid.
- vi. and vice versa
- 35 27. A method according to claim 1 or 25, comprising multiple subsequent steps of

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- vii. introducing into the vessel at least one of said reactant(s) and/or precursor(s) and/or initiator(s) and/or catalyst(s) for said chemical reaction
- 5 viii. subsequently introducing into the vessel one or more substances dissolved and/or dispersed in at least one fluid.
- ix. and vice versa
- 10 28. A method according to claim 1 or 26, wherein said material in said one or more section(s) is capable of adsorbing at least one of said reactant(s) and/or precursor(s) and/or catalyst(s) on said material.
- 29. A method according to claim 1 or 27, wherein said reactant(s) and/or precursor(s)
- 15 and/or catalyst(s) is/are adsorbed substantially in a monolayer of said material.
- 30. A method according to any of the claims 1, 20-29, wherein the time for said chemical reaction(s) is less than 24 hours, such as less than 12 hours and preferable less than 8 hours, and even more preferable less than 4 hours.
- 20 31. A method according to any of the claims 1, 20-29, wherein the time for said chemical reaction(s) is maximum 2 hours, such as maximum 1 hour, preferably less than 30 minutes and even more preferably less than 15 minutes.
- 25 32. A method according to any of the preceding claims, wherein said material present in said one more section(s) of said vessel, provides a distributing effect of said fluid(s) being introduced into said vessel.
- 33. A method according to claim 32, wherein the fraction of the total volume comprised
- 30 by the said material in said one or more section(s) is up to 70 %, such as up to 50 %, preferably up to 30 %, and even more preferably up to 20 %.
- 34. A method according to any of the preceding claims, wherein said material present in said one or more section(s) of said vessel, comprises additional nucleation sites.
- 35 35. A method according to any of the preceding claims, wherein said material present in said one or more section(s) of said vessel, provides a seeding effect.

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36. A method according to claim 34, wherein the number of nucleation sites is further increased by introducing an ultrasound and/or a vibrating surface effect.

37. A method according to any of the claims 34-36, wherein said material present in
5 said one or more sections is a template for forming said primary particles into a specific shape, size, structure or phase.

38. A method according to claim 37, wherein said primary particles being produced is a least partly crystalline.

10

39. A method according to claim 36-38, wherein said seeding effect is at least partly provided by seed particles.

40. A method according to claim 39, wherein said seed particles is in a fluidised or
15 suspended state.

41. A method according to the claims 39-40, wherein said seed particles comprises the same material as said primary particles.

20 42. A method according to any of the claims 32-41, wherein said material present in said one or more sections comprises a porous structure such as a sheet, a spongeous or a grid structure.

43. A method according to any of the claims 32-41, wherein said material present in
25 said one or more sections is a fibrous material.

44. A method according to claims 42-43, wherein said material present in said one or more sections has/have a hydrophilic surface.

30 45. A method according to claims 42-43, wherein said material present in said one or more section has/have a hydrophobic surface.

46. A method according to any of the claims 42-45, wherein said material is a polymer material.

35

47. A method according to claim 46, wherein the material in the vessel is a polymer or elastomer such as selected from the group consisting of polyethylene, polypropylene, polystyrene, polyesters, polyethylene terephthalate, polyvinyl chloride, polyvinyl acetates, polyoxymethylene, polyacryloamide, polycarbonate, polyamides, polyurethane, copolymers

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thereof, chlorinated products thereof, rubbers and chlorinated rubber, silicone rubbers, butadiene rubbers, styrene-butadiene-rubbers, isoprene polymers, vulcanised fluororubbers, silicone rubbers.

5 48. A method according to any of the claims 47, wherein said polymer material is polypropylene.

49. A method according to claim 46, wherein said fibrous material is an elastic material.

10

50. A method according to any of the claims 42-45, wherein said material is a ceramic material.

51. A method according to claim 50, wherein said material is a glass wool such as
15 quarts wool.

52. A method according to claim 50, wherein said material is an aerogel.

53. A method according to claim 52 comprising the steps of
20 - producing said aerogel material by a sol-gel reaction in an organic solvent
- removing said organic solvent by extraction in supercritical CO₂
- drying at least partly said aerogel
- forming said primary particles on the surface of said aerogel by a method
according to any of the preceding claims.

25

54. A method according to any of the claims 42-53, wherein the specific surface area (m²/m³) of said material in said sections is above 500 m²/m³, such as 1000 m²/m³, such as above 10.000 m²/m³, and preferably above 50.000 m²/m³ such as above 100.000 m²/m³.

30 55. A method according to claim 54, wherein said material is a filtration medium.

56. A method according to any of the claims 54-55, wherein said material present in said one or more sections in the vessel further comprises at least one hollow tubular member comprising an inner and an outer surface, and having at least one end
35 communicating with the outside of said vessel, at least part of said hollow tubular member(s) comprising a membrane, at least one of the fluids being introduced into the vessel through the membrane.

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57. A method according to claim 56, wherein said hollow tubular member(s) is blocked in the one end being inside the vessel.

58. A method according to claim 57, wherein both ends of said hollow tubular member(s) are open and are communicating with the outside of said vessel.

59. A method according to claim 58, wherein said membrane contains at least one dense layer.

60. A method according to any of the preceding claims 56-59, wherein said membrane is a porous membrane, preferably being a membrane having pores within a range of 0,001-100 micron, such as a pores in the range 0,01-10 micron, and preferably being a membrane having pores within the range of 0,01- 0,1 micron.

61. A method according to any of the claims 56-60, wherein the inlet end of said hollow tubular member(s) is communicating with an inlet plenum, wherein an antisolvent is introduced, at least part of said antisolvent permeating said membrane surface and providing a controlled and uniform distribution of said antisolvent in said vessel.

62. A method according to any of the claims 60-61, wherein at least one of said fluids containing dissolved substances is introduced into the vessel through said hollow tubular member(s).

63. A method according to any of the claims 61 or 62, wherein at least two fluid streams are introduced into the vessel through two separate sets of hollow tubular member(s) each having an inlet plenum communicating with the outside of the vessel.

64. A method according to any of the claims 62-63, comprising contacting at least part of the inner surface of said hollow tubular member(s) with a first fluid, and contacting at least part of the outer surface(s) of said hollow tubular member(s) with a second fluid, a least part of said first fluid permeating through at least part the wall(s) of said hollow tubular member(s), and mixes with said second fluid substantially at the outer surface(s) of said hollow tubular member(s).

65. A method according to claim 64, wherein a micro-emulsion of said first fluid in said second fluid is formed.

67. A method according to any of the claims 62-65, wherein one of the fluids is water or contains water.

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67. A method according to claim 66, wherein said water or water mixture contains one or more substances preferably being dissolved or dispersed therein.

5 68. A method according to any of the claims 62-67, wherein one of the fluids is CO₂ or an oil.

69. A method according to claim 68, wherein said fluid further comprises one or more surfactants.

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70. A method according to any of the claims 65-69, wherein said microemulsion formed comprises a water core.

71. A method according to claim 70, wherein said water core comprises dissolved
15 and/or dispersed substances.

72. A method according to the claims 70-71, wherein the diameter of said water core in said emulsions formed is in the range 0,5-15 times the diameter of the pores of the membrane part of said hollow tubular member(s) contacting said second fluid, such as in
20 the range 1-10 times the diameter of the pores of the membrane part of said hollow tubular member(s) contacting said second fluid, and preferably in the range 2-4 times the diameter of the pores of the membrane part of said hollow tubular member(s) contacting said second fluid.

25 73. A method according to any of the claims 66-72, wherein said material present in said one or more sections, comprises two sets of hollow tubular member(s), both sets of said hollow tubular member(s) comprising a plenum and an outlet plenum communicating with the outside of said vessel, and wherein two different fluids can be contacted with the inner surface of said hollow tubular member(s), and where two different emulsion(s) of
30 said fluids in said fluid contacting the outer surface of said hollow tubular member(s) are formed.

74. A method according to any of the claims 66-73, wherein a reaction is occurring in said microemulsion(s).

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75. A method according to any of the claims 66-74, wherein said fluid containing said microemulsion(s) are used to dissolve and/or extract substances outside of said vessel.

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76. A method according to any of the preceding claims, wherein said vessel is operating at a substantially constant pressure, such as operating continuously at a constant pressure during pre-selected time intervals.

5 77. A method according to any of the preceding claims, wherein said vessel is at a substantially constant pressure at more than one subsequent pre-selected pressure levels.

78. A method according to any the preceding claims, comprising re-circulating in at least part time of the method at least part of a fluid mixture present in the vessel, the re-
10 circulating comprising:

- withdrawing from the vessel at least a part of a fluid from the vessel and feeding it to a re-circulation loop and subsequently feeding the fluid back to the vessel.

15 79. A method according to claim 78, further comprising the step of controlling the temperature of the fluid in the re-circulation loop.

80. A method according to any of the claims 78-79, wherein heat is added and/or extracted from the fluid in the re-circulation loop.

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81. A method according to any of the claims 78-80, wherein one or more reactant is added and/or extracted from the fluid in the re-circulation loop.

82. A method according to 80, wherein at least one of the reactants is an alcohol an
25 alkoxide and/or water.

83. A method according to claim 81, wherein a metal- or semi-metal alkoxid are produced in-situ in the process prior to being introduced to said vessel by said fluid.

30 84. A method according to any of the preceding claims, wherein the method comprises controlling the temperature- and/or pressure- and/or density- and/or concentration profiles within the vessel.

85. A method according to any of the preceding claims, wherein the temperature
35 profile within the vessel is controlled by controlling the temperature and flow rate of at least one fluid flowing inside said hollow tubular members.

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86. A method according to any of the preceding claims, wherein said material present in said one or more sections in vessel with said precipitated primary particles thereon comprises the final product.

5 87. A method according to claim 86, wherein said product comprises primary particles deposited on a carrier film such as a tape cast.

88. A method according to claim 86, wherein said primary particles on said surface of said material constitutes a film or a coating.

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89. A method according to claim 87, wherein said film or coating has one or more layer(s) each layer having a layer thickness of up to 1 micron, such as a layer thickness below 500 nanometer, preferable a layer thickness below 250 nanometer such as a layer thickness below 100 nanometer. Even more preferable the layer thickness of said film is
15 below 50 nanometer, such as a layer thickness below 30 nanometer.

90. A method according to any of the claims 87-88, wherein said coating comprises multiple layers.

20 91. A method according to claim 89, where the layers comprises different materials.

92. A method according to any of the claims 87-90, wherein said product is further subjected to an annealing process.

25 93. A method according to claim 91, wherein said annealing is performed by microwaves.

94. A method according to claim 86, wherein said primary particles are deposited on the surface of said material in the form of small clusters of individual particles, and
30 preferably as individual particles.

95. A method according to claim 93, wherein said clusters comprises up to 100 atoms , such as up to 50 atoms, and preferably less than 10 atoms, and even more preferably less than 5 atoms.

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96. A method according to any of the claims 93-94, wherein said clusters or individual particles on the surface of said material are deposited as quantum dots.

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97. A method according to any of the preceding claims, wherein said primary particles precipitated on said surface of the material present in said one or more section(s) are removed from said material as a powder.

5 98. A method according to claim 96, wherein said powder consists of weakly bounded agglomerates of primary particles.

99. A method according to claim 97, wherein said powder has weakly bounded agglomerates of a size of maximum 10 micron, such as up to 5 micron, and preferably up
10 to 1 micron such as up to 500 nanometer.

100. A method according to any of the claims 96-98, wherein said powder is removed from said material by introducing a vibrating effect and/or an acoustic effect such as ultrasound waves and/or by back flushing and/or by applying an pressure pulse effect.
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101. A method according to claim 99, wherein said vibrating effect is generated by piezoelectric means.

102. A method according to claim 99, wherein said vibrating effect is generated by a
20 magneto-restrictive means.

103. A method according to any of the claims 96-101, wherein said powder is removed from said material while said material is within the vessel.

25 104. A method according to claim 102, wherein said removal of said powder are performed according to a back flush or back pulse or a back chock technique.

105. A method according to any of the claims 102-103, wherein said removed powder are withdrawn from the vessel by flushing with a fluid or fluid mixture present in the
30 vessel.

106. A method according to claim 104, wherein said fluid containing said formed powder is fed into a second vessel containing a liquid.

35 107. A method according to any of the claims 104-105, wherein said fluid containing said formed powder is expanded into said liquid thereby providing said formed powder material as a dispersion in said liquid.

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108. A method according to claim 104, wherein said fluid containing said formed powder is fed to a bag filter or ceramic filter for separation of said formed powder material from the fluid.

5 109. A method according to claim 104, wherein said fluid containing said formed particulate material is fed to a membrane separation device.

110. A method according to claim 104, wherein said formed powder contained in said fluid is deposited on to a second solid in a second vessel.

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111. A method according to claim 109, wherein said deposition is performed by spraying.

112. A method according to any of the preceding claims, wherein said precipitated
15 primary particles on said material are exposed to one or more coating or encapsulation steps and/or reaction steps with one or more materials.

113. A method according to claim 111, wherein said coating or encapsulation step(s) is performed within the vessel.

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114. A method according to any of the claims 111-112, wherein said coating or encapsulation step(s) is performed at least partly during harvesting/removing said particles from said material.

25 115. A method according any of the claims 111-113, wherein said coating or a further coating is performed in a second vessel.

116. A method according to any of the preceding claims, wherein said nanomaterial production process involves one of the following processes: RESS (rapid expansion of
30 supercritical solutions), GAS (Gas Antisolvent), SAS (solvent Anti Solvent), SEDS (Solution Enhanced Dispersion by supercritical fluid), PCA (Precipitation with Compressed Antisolvent), PGSS (Precipitation from Gas-saturated Solutions) and variations thereof.

117. A method according to any of the preceding claims, wherein said primary particles
35 comprises an electro-ceramic material.

118. A method according to any of the preceding claims, wherein said primary particles comprises a semi-conducting material.

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119. A method according to any of the preceding claims, wherein said primary particles comprises a magnetic, ferromagnetic, paramagnetic, or superparamagnetic material.

120. A method according to any of the preceding claims, wherein said primary particles
5 comprises a core-shell structure.

121. A method according to any of the preceding claims, wherein said core comprises a magnetic or ferro magnetic core.

10 122. A method according to any of the preceding claims, wherein said primary particles comprises a piezoelectrical material.

123. A method according to claim 121, wherein said piezoelectrical material comprises lead zirconate titanate, $\text{Pb}(\text{Zr}_{0.52}\text{Ti}_{0.48})\text{O}_3$

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124. A method according to any of the preceding claims, wherein said primary particles comprises oxide(s), oxyhydroxide(s), hydroxide(s) such as metal oxide(s), semi-metal oxide(s), metal oxyhydroxide(s), semi-metal oxyhydroxide(s), metal hydroxide(s), semi-metal hydroxides and combinations thereof.

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125. A method according to any of the claims 116-123, wherein said oxides comprises oxides of one or more of the following elements: Al, Si, Ti, Zr, Zn, Fe, Ni, Co, Ce, Ge, Ba, Sr, W, La, Ta, Y, Mn, V, Bi, Sn, Te, Se, Ga, Be, Pb, Cr, Mg, Ca, Li, Ag, Au, Pt, Pd, Cd, Mo, Eu and combinations thereof.

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126. A method according to claim 124, wherein said oxides is selected among silica, alumina, zirconia, titania and combinations thereof.

127. A method according to any of the claims, wherein said primary particles comprises
30 carbide(s) and/or nitride(s).

128. A method according to any of the claims 123-124, wherein said metal or semi-oxide(s) is/are precursor(s) for a thermoelectric material.

35 129. A method according to claim 127, wherein said precursor(s) is/are reduced in the process by adding a reducing agent to form a thermoelectric material.

130. A method according to claim 128, wherein said thermoelectric material formed comprises a clathrate.

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131. A method according to any of the claims 127-129, said thermoelectric material comprises one or more of the elements: Ba, Bi, Te, Se, Zn, Sn, Sr, Ga, Ge, Pb, Cd, Sb, Ag, Si and combinations thereof.

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132. A method according to claim 130, further comprising doping said primary particles of said thermoelectric material with metals and/or semi-metals.

133. A method according to claim 131, wherein said thermoelectrical material have a thermal conductivity at temperatures above 20 C of maximum 10 watts per meter Kelvin, such as maximum 5 watts per meter Kelvin, preferably maximum 3 watts per meter Kelvin such as maximum 1,5 watts per meter Kelvin, and even more preferably a heat conductivity of maximum 1 watt per meter Kelvin.

134. A method according to any of the preceding claims, wherein said primary particles comprises one or more pharmaceutical and/or biological material(s).

135. A method according to any of the preceding claims, wherein said primary particles deposited on the surface of said surface provides an antibacterial effect of said surface.

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136. A method according to any of the preceding claims, wherein said material comprises a high surface area material.

137. An apparatus comprising one or more of the means disclosed in any of the preceding claims and being adapted to carry out the method according to any of the preceding claims.

138. A product obtainable according to a method in any of the preceding claims.

139. A tape cast for tape casting, comprising primary particles deposited on a carrier film, wherein said primary particles have:

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- a. an average diameter of less than 100 nanometer such as an average diameter of less than 30 nanometer, preferably an average diameter of smaller than 20 nanometer and even more preferable an average diameter below 15 nanometer such as below 10 nanometer.
- b. a narrow size distribution around the average diameter characterized by having a maximum standard deviation of said distribution of maximum 20 nanometer, such as maximum 10 nanometer, and preferably less than 5

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nanometer.

140. A tape cast according to claim claim 138, suitable for production of a ceramic material by tape casting.
- 5 141. A tape cast according to claim 139, wherein the sintering temperature is maximum 1100 C, such as maximum 1000 C, and preferable maximum 900 C, and even more preferably maximum 800 C, such as 700 C.
- 10 142. A tape cast according to claim 140, wherein the material being produced is a piezomotor produced from lead zirconate titanate tape cast.
143. An item having a hard nanocrystalline coating comprising primary particles of Al_2O_3 and ZrO_2 according to any of the preceding claims, wherein said coating has a hardness of
15 at least 10 GPA, such as a hardness of at least 15 GPA, and preferably above 20 GPA, and even more preferably a hardness of at least 25 GPA.
144. An item having a hard nanocrystalline coating comprising primary particles of Al_2O_3 and ZrO_2 according to any of the preceding claims, wherein said coating has a scratch and
20 wear resistance of at least 30 N, such as a scratch and wear resistance of at least 35 N, preferably a scratch and wear resistance of at least 40 N, and even more preferably a scratch resistance of at least 45 N.
145. An item having a hard nanocrystalline coating according to any of the claims 142-
25 143, further comprising primary particles of ZnO .
146. A hard nanocrystalline coating according to any of the claims 142-144, wherein said coating is applied to a polymer or a glass material.
- 30 147. A mechanical part with a hard nanocrystalline coating according to any of the claims 142-145, wherein said coating is applied to the surface of said material.